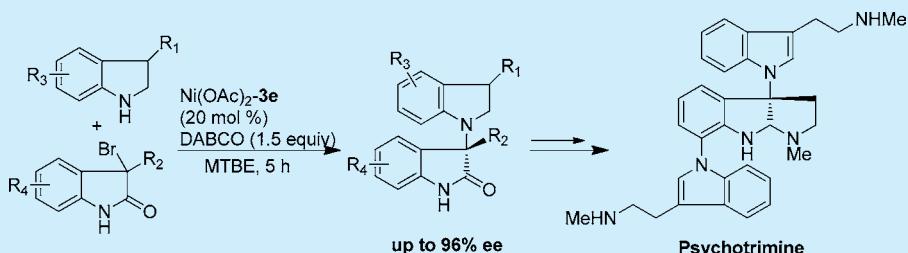


Construction of the N1–C3 Linkage Stereogenic Centers by Catalytic Asymmetric Amination Reaction of 3-Bromooxindoles with Indolines

Hailong Zhang,[†] Hong Kang,[†] Liang Hong, Weiping Dong, Guolin Li, Xin Zheng, and Rui Wang*

School of Life Sciences, Key Laboratory of Preclinical Study for New Drug of Gansu Province, Lanzhou University Lanzhou 730000, P.R. China

Supporting Information



ABSTRACT: The catalytic asymmetric amination reaction of 3-bromooxindoles with indolines for the construction of the N1–C3 linkage stereogenic centers has been realized for the first time. Moreover, the racemic substrates (3-substituted indolines) were also applicable under the same chiral conditions. The newly developed method conveniently led to a formal synthesis of (+)-psychotrimine.

The indole-derived pyrroloindolines widely exist in a large number of natural products and pharmacologically important compounds.^{1,2} However, a small collection of molecules, including chaetomin,³ kapakahines,⁴ psychotrimine,⁵ and psychotetramine,^{5f} are linked through N1–C3 bonds⁷ (Figure 1). These indole alkaloids show unique structures and

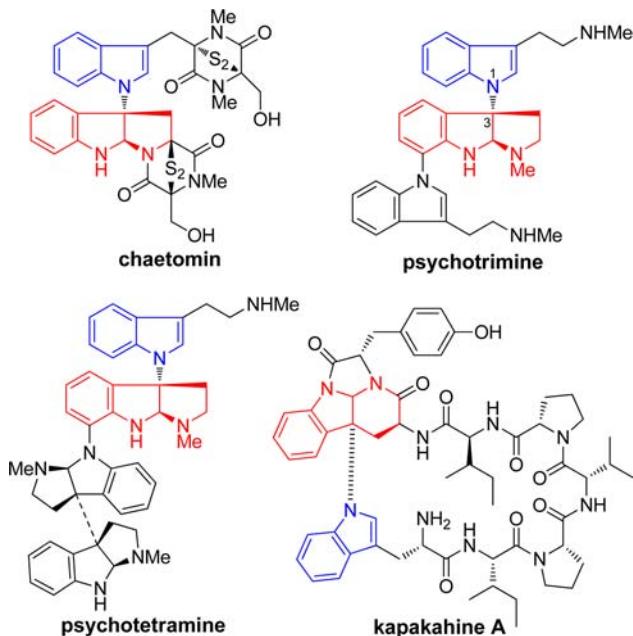


Figure 1. Representative members of the N1–C3 linkage indole alkaloids.

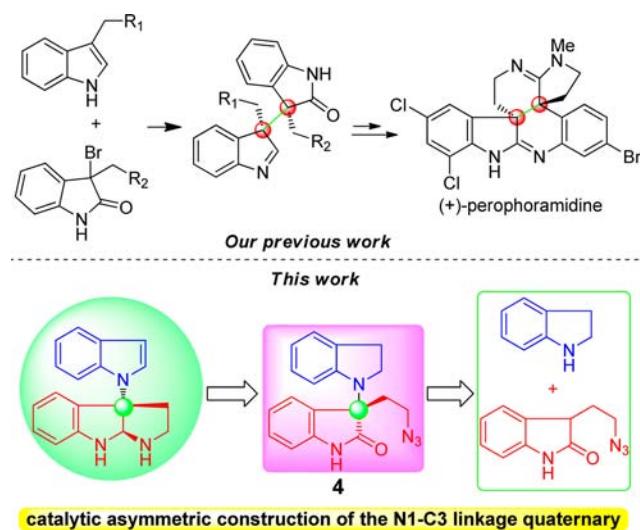
interesting bioactivity profiles. For example, psychotrimine, a trimeric-tryptamine-related alkaloid, was isolated from the leaves of the plant *Psychotria rostrata* by Takayama and co-workers in 2004^{5a} and exhibits potent antitumor activity against colon and lung cancers^{5e} and antibacterial activity against Gram-positive bacteria.⁶ The unique structure and interesting biological activity of psychotrimine have attracted considerable attention from synthetic chemists worldwide.^{5b–f} Takayama et al. employed an asymmetric Ireland–Claisen rearrangement as a key step for the first enantioselective total synthesis of (+)-psychotrimine.^{5e} An elegant total synthesis of (+)-psychotrimine was achieved by Baran and co-workers, through the direct aniline–indole coupling of *o*-iodoaniline with 7-bromo-D-tryptophan derivative.^{5f}

The N1–C3 linkage stereogenic center is the pronounced feature in these natural products. From a synthetic point of view, several members of the indole alkaloids contain the N1–C3 linkage stereogenic center that could potentially be accessed via 3-aminoxindole⁸ 4 (Scheme 1). Therefore, the catalytic asymmetric construction of the N1–C3 linkage stereogenic center in 4 would be the critical element in the divergent synthesis of these indole alkaloids. Unsurprisingly, the methods to construct this type of C–N bond directly in an asymmetric and catalytic fashion are rare.^{9–11} Recently, we disclosed an asymmetric alkylation reaction of 3-bromooxindoles with 3-substituted indoles with high diastereoselectivity and excellent enantioselectivity by using a nickel(II) catalyst. We found that the enantioselectivity was controlled by the chiral electrophile

Received: March 10, 2014

Published: April 11, 2014

Scheme 1. Strategically Diversity-Oriented Retrosynthetic Analysis

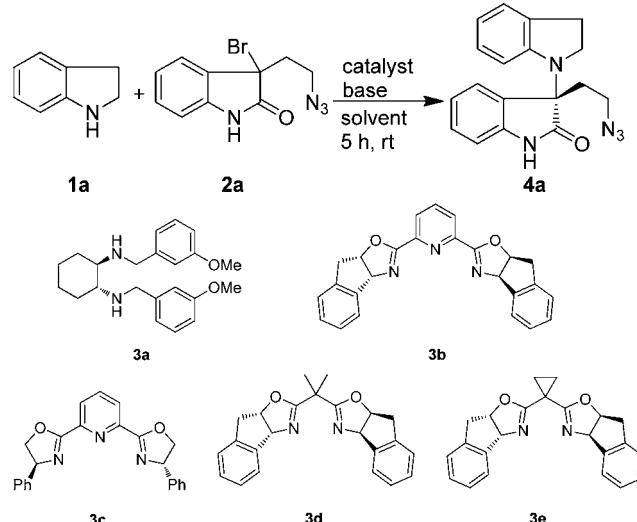


(indol-2-one) complex induced by a chiral nickel(II) catalyst.¹² Herein, we developed an efficient catalytic asymmetric amination reaction of 3-bromooxindoles¹³ with indolines for the construction of the N1–C3 linkage stereogenic centers. This method also could be convenient for the synthesis of (+)-psychotrimine.

In our initial experiments, we investigated an examination of the reaction between 3-bromooxindole **2a** and indoline **1a** in the presence of chiral nickel(II) catalysts. As shown in Table 1, the 1:1 $\text{Ni}(\text{OAc})_2\text{-}3\mathbf{a}$ complex (20 mol %, THF, rt, 5 h) combined with K_3PO_4 , which we have been developing in a previous study,¹² afforded the desired product **4a** in excellent yield, but with low enantioselectivity (Table 1, entry 1). However, the replacement of **3a** with other ligands, such as **3b**, **3c**, **3d**, or **3e**, gave no significant improvement to the enantioselectivity (Table 1, entries 2–5). Then, several bases were surveyed. To our great delight, when DABCO was used as a base, the enantioselectivity was enhanced (75% ee, Table 1, entry 9). Subsequent screening of solvents revealed that MTBE appeared to be the most suitable reaction media for high enantioselectivity (up to 87% ee; Table 1, entries 14). Finally, we found that lowering the amount of base results in a slightly enhanced enantioselectivity at 90% ee (Table 1, entry 15).

Having established the optimal reaction conditions, the scope of substrates for the catalytic asymmetric amination reaction of 3-bromooxindoles with indolines was then studied (Scheme 2). On the whole, it was found that a broad range of 3-bromooxindoles and indolines could readily participate in this reaction. We first investigated a variety of indolines under the optimal reaction conditions. Indolines with electron-withdrawing groups gave adducts with high enantioselectivity (88–96% ee) and high yield (71–88%), but indolines with electron-donating groups experienced lower enantioselectivity (**4c**, 74% ee; **4d**, 61% ee). Furthermore, the scope of 3-bromooxindoles was surveyed. Substitution of the bromooxindole with a methoxy group at C5 produced the indoline amination product with good results (**4k**). In contrast, only moderate selectivity and yield were observed with substrate **2l**, which contains an electron-withdrawing C5-Cl on the bromooxindole core. In addition, bromooxindoles bearing different 3-substitution/alkyl chains could participate in this

Table 1. Optimization of the Reaction^a



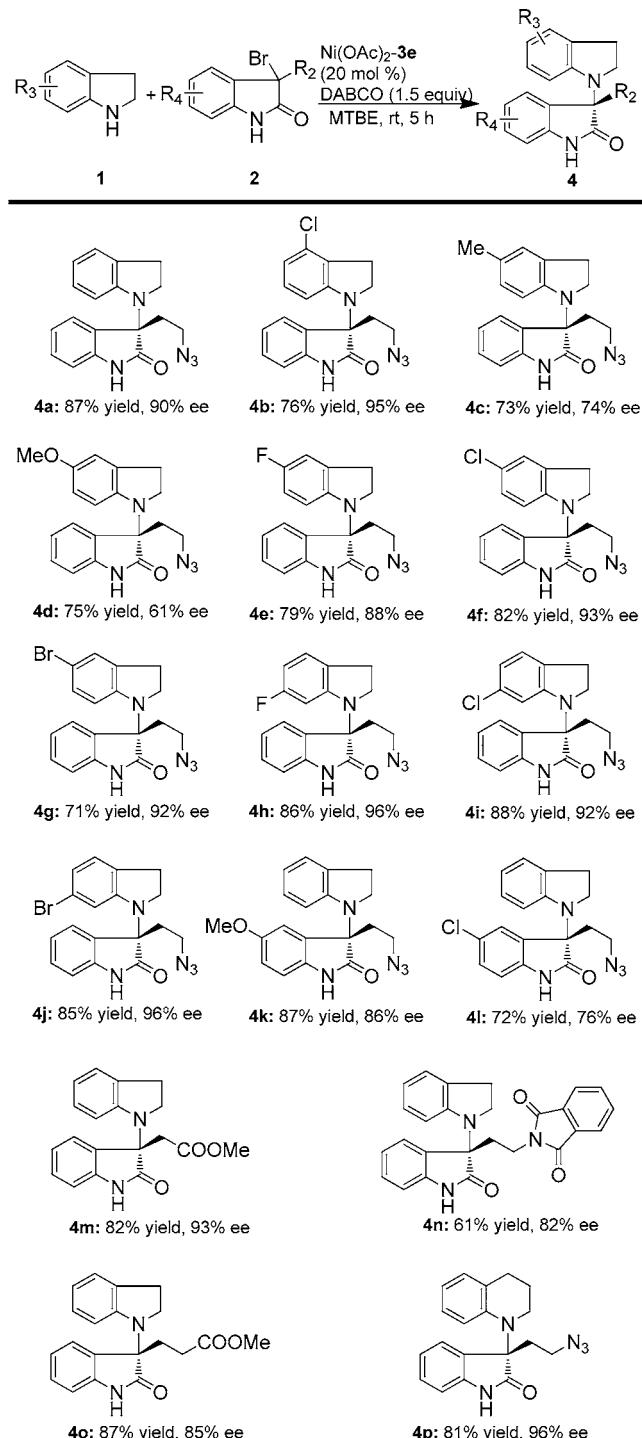
^aUnless otherwise specified, the reaction was carried out with **1a** (0.12 mmol), **2a** (0.1 mmol), and base (0.2 mmol) in the presence of catalyst (0.02 mmol) and solvent (1.5 mL). ^bIsolated yield.

^cDetermined by chiral HPLC on a Chiraldak OD column. ^dDABCO (0.15 mmol) was added. DBU = 1,8-diazabicyclo[5.4.0]undec-7-ene, DIPEA = *N,N*-diisopropylethylamine, DABCO = triethylenediamine, DME = 1,2-dimethoxyethane, CPME = cyclopentyl methyl ether, MTB E = methyl *tert*-butyl ether.

reaction (**4m–o**). To our delight, in addition to indolines, the tetrahydroquinoline was also suitable for the reaction to give the respective adduct with excellent results (**4p**).

On the basis of our recent research,¹² we surmised that racemic 3-substituted indolines could amine 3-bromooxindoles stereoselectively. As shown in Scheme 3, several 3-substituted indolines can participate in the process to give the desired amination adducts **4** with moderate yield. The ee value was determined by the conversion of indoline **4** into indole **5** with DDQ oxidation. Notably, the product **5d** was obtained in moderate yield with excellent stereoselectivity when the tryptophan-derived indoline was employed. The potential utility of this reaction was supported by the relationship between Chaetomin's skeleton and **5d**.

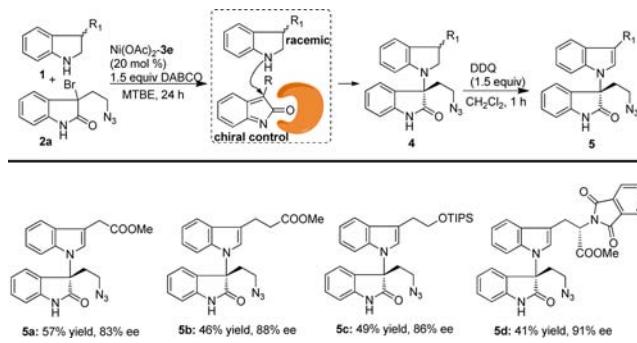
The synthetic utility of this strategy was demonstrated by the aminated product **4a**,^{5e} which has been achieved in 12 chemical steps by Takayama et al., as a key intermediate in the enantioselective synthesis of (+)-psychotrimine (Scheme 4).

Scheme 2. Scope of the Reaction^a

^aUnless otherwise specified, the reaction was carried out with **1** (0.12 mmol), **2** (0.1 mmol), and base (0.15 mmol) in the presence of a catalyst (0.02 mmol) and solvent (1.5 mL). Isolated yield. The ee values were determined by HPLC on a chiral phase (Chiralcel column).

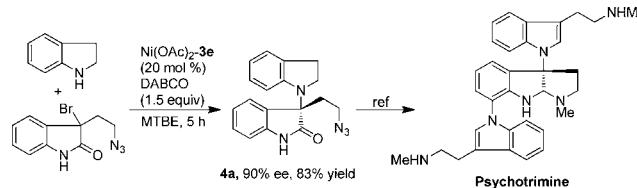
Here, we could expediently obtain a series of **4a** analogues by the direct catalytic asymmetric amination reaction of 3-bromooxindoles with indolines, and the chemical diversity of (+)-psychotrimine may be accessed via this method.

In conclusion, we have developed a mild enantioselective nickel(II)-catalyzed amination reaction of 3-bromooxindoles

Scheme 3. Scope of the Racemic 3-Substituted Indolines^a

^aThe reaction was performed on 0.1 mmol scale with **1** (1.2 equiv) and **2a** (1.0 equiv). DDQ = 2,3-dichloro-5,6-dicyanobenzoquinone.

Scheme 4. Formal Synthesis of Psychotrimine



with indolines for the construction of the N1–C3 linkage quaternary stereogenic centers. Notably, the racemic substrates (3-substituted indolines) could also participate in the reaction with high stereoselectivity. Furthermore, this method facilitated access to cytotoxic (+)-psychotrimine. Further application of this method to this class of alkaloids is underway.

ASSOCIATED CONTENT

S Supporting Information

General methods, experimental and HPLC data, and ¹H and ¹³C NMR spectra. This material is available free of charge via the Internet at <http://pubs.acs.org>.

AUTHOR INFORMATION

Corresponding Author

*E-mail: wangrui@lzu.edu.cn.

Author Contributions

[†]These authors contributed equally.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We gratefully acknowledge financial support from NSFC (91213302, 21272102, 21202071) and the National S&T Major Project of China (2012ZX09504001-003).

REFERENCES

- (a) Anthoni, U.; Christophersen, C.; Nielsen, P. H. Naturally Occurring Cyclotryptophans and Cyclotryptamines. In *Alkaloids: Chemical Biological & Perspectives*; Pelletier, S. W., Ed.; Pergamon: Oxford, U.K., 1999; Vol. 13, pp 163–236. (b) Julian, P. L.; Pikl, J. *J. Am. Chem. Soc.* **1933**, *55*, 2105. (c) Julian, P. L.; Pikl, J.; Boggess, D. *J. Am. Chem. Soc.* **1934**, *56*, 1797. (d) Julian, P. L.; Pikl, J. *J. Am. Chem. Soc.* **1935**, *57*, 755. (e) Cordell, G. A.; Saxton, J. E. In *The Alkaloids*; Elsevier: Amsterdam, 1981; Vol. 20, p 10. (f) Steven, A.; Overman, L. E. *Angew. Chem., Int. Ed.* **2007**, *46*, 5488. (g) Schmidt, M. A.;

Movassagh, M. *Synlett* **2008**, 3, 313. (h) Ruiz-Sanchis, P.; Savina, S. A.; Albericio, F.; Alvarez, M. *Chem.—Eur. J.* **2011**, 17, 1388.

(2) For recent reports of the synthesis of pyrrolidinoindoline alkaloids, see: (a) Dounay, A. B.; Overman, L. E. *Chem. Rev.* **2003**, 103, 2945. (b) Kodanko, J. J.; Overman, L. E. *Angew. Chem., Int. Ed.* **2003**, 42, 2528. (c) Austin, J. F.; Kim, S.-G.; Sinz, C. J.; Xiao, W.-J.; MacMillan, D. W. C. *Proc. Natl. Acad. Sci. U.S.A.* **2004**, 101, 5482. (d) Menozzi, C.; Dalko, P. I.; Cossy, J. *Chem. Commun.* **2006**, 42, 4638. (e) Movassagh, M.; Schmidt, M. A. *Angew. Chem., Int. Ed.* **2007**, 46, 3725. (f) Kim, J.; Ashenhurst, J. A.; Movassagh, M. *Science* **2009**, 324, 238. (g) Repka, L. M.; Reisman, S. E. *J. Org. Chem.* **2013**, 78, 12314. (h) Kieffer, M. E.; Chuang, K. V.; Reisman, S. E. *J. Am. Chem. Soc.* **2013**, 135, 5557. (i) Repka, L. M.; Ni, J.; Reisman, S. E. *J. Am. Chem. Soc.* **2010**, 132, 14418.

(3) (a) Waksman, S. A.; Bugie, E. *J. Bacteriol.* **1944**, 48, 527. (b) Geiger, W. B.; Conn, J. E.; Waksman, S. A. *J. Bacteriol.* **1944**, 48, 531. (c) McInnes, A. G.; Taylor, A.; Walter, J. A. *J. Am. Chem. Soc.* **1976**, 98, 6741. (d) Brewer, D.; McInnes, A. G.; Smith, D. G.; Taylor, A.; Walter, J. A.; Loosli, H. R.; Kis, Z. L. *J. Chem. Soc., Perkin Trans. 1* **1978**, 10, 1248.

(4) (a) Nakao, Y.; Yeung, B. K. S.; Yoshida, W. Y.; Scheuer, P. J.; Kelly-Borges, M. *J. Am. Chem. Soc.* **1995**, 117, 8271. (b) Yeung, B. K. S.; Nakao, Y.; Kinnel, R. B.; Carney, J. R.; Yoshida, W. Y.; Scheuer, P. J.; Kelly-Borges, M. *J. Org. Chem.* **1996**, 61, 7168. (c) Nakao, Y.; Kuo, J.; Yoshida, W. Y.; Kelly, M.; Scheuer, P. J. *Org. Lett.* **2003**, 5, 1387.

(5) For the isolation, see: (a) Takayama, H.; Mori, I.; Kitajima, M.; Aimi, N.; Lajis, N. H. *Org. Lett.* **2004**, 6, 2945. For the racemic syntheses, see: (b) Matsuda, Y.; Kitajima, M.; Takayama, H. *Org. Lett.* **2008**, 10, 125. (c) Newhouse, T.; Baran, P. S. *J. Am. Chem. Soc.* **2008**, 130, 10886. (d) Newhouse, T.; Lewis, C. A.; Eastman, K. J.; Baran, P. S. *J. Am. Chem. Soc.* **2010**, 132, 7119. For the enantioselective syntheses, see: (e) Takahashi, N.; Ito, T.; Matsuda, Y.; Kogure, N.; Kiyajima, M.; Takayama, H. *Chem. Commun.* **2010**, 46, 2501. (f) Foo, K.; Newhouse, T. R.; Mori, I.; Takayama, H.; Baran, P. S. *Angew. Chem., Int. Ed.* **2011**, 50, 2716. (g) Araki, T.; Ozawa, T.; Yokoe, H.; Kanematsu, M.; Yoshida, M.; Shishido, K. *Org. Lett.* **2013**, 15, 200.

(6) Schallenger, M. A.; Newhouse, T.; Baran, P. S.; Romesberg, F. E. *J. Antibiot.* **2010**, 63, 685.

(7) (a) Kung, A. L.; Zabludoff, S. D.; France, D. S.; Freedman, S. J.; Tanner, E. A.; Vieira, A.; Cornell-Kennon, S.; Lee, J.; Wang, B.; Wang, J.; Memmert, J.; Naegeli, H.-U.; Petersen, F.; Eck, M. J.; Bair, K. W.; Wood, A. W.; Livingston, D. M. *Cancer Cell* **2004**, 6, 33. (b) Masashi, H.; Mitsuyoshi, S.; Hiroshi, M.; Yuji, S. *Tennen Yuki Kagobutsu Toronkai Koen Yoshishu* **1994**, 36, 557. (c) Li, G.-Y.; Li, B.-G.; Yang, T.; Yan, J.-F.; Liu, G.-Y.; Zhang, G.-L. *J. Nat. Prod.* **2006**, 69, 1374. (d) Ding, G.; Jiang, L.; Guo, L.; Chen, X.; Zhang, H.; Che, Y. *J. Nat. Prod.* **2008**, 71, 1861.

(8) For 3-aminooxindole, see: (a) Emura, T.; Esaki, T.; Tachibana, K.; Shimizu, M. *J. Org. Chem.* **2006**, 71, 8559. (b) Bergonzini, G.; Melchiorre, P. *Angew. Chem., Int. Ed.* **2012**, 51, 971. (c) Noole, A.; Järving, I.; Werner, F.; Lopp, M.; Malkov, A.; Kanger, T. *Org. Lett.* **2012**, 14, 4922. (d) Li, J.; Du, T.; Zhang, G.; Peng, Y. *Chem. Commun.* **2013**, 49, 1330. (e) Zhou, F.; Zeng, X.-P.; Wang, C.; Zhao, X.-L.; Zhou, J. *Chem. Commun.* **2013**, 49, 2022. (f) Hara, N.; Nakamura, S.; Sano, M.; Tamura, R.; Funahashi, Y.; Shibata, N. *Chem.—Eur. J.* **2012**, 18, 9276. (g) Chen, D.; Xu, M.-H. *Chem. Commun.* **2013**, 49, 1327. (h) Mouri, S.; Chen, Z.; Mitsunuma, H.; Furutachi, M.; Matsunaga, S.; Shibusaki, M. *J. Am. Chem. Soc.* **2010**, 132, 1255. (i) Bui, T.; Hernández-Torres, G.; Milite, C.; Barbas, C. F., III. *Org. Lett.* **2010**, 12, 5696. (j) Ren, L.; Lian, X.-L.; Gong, L.-Z. *Chem.—Eur. J.* **2013**, 19, 3315. (k) Shen, K.; Liu, X.; Wang, G.; Lin, L.; Feng, X. *Angew. Chem., Int. Ed.* **2011**, 50, 4684.

(9) For racemic examples, see: (a) Espejo, V. R.; Rainier, J. D. *J. Am. Chem. Soc.* **2008**, 130, 12894. (b) Beaumont, S.; Pons, V.; Retailleau, P.; Dodd, R. H.; Dauban, P. *Angew. Chem., Int. Ed.* **2010**, 49, 1634. (c) Espejo, V. R.; Li, X.-B.; Rainier, J. D. *J. Am. Chem. Soc.* **2010**, 132, 8282.

(10) For chiral-auxiliary-controlled examples, see: (a) Movassagh, M.; Ahmad, O. K.; Lathrop, S. P. *J. Am. Chem. Soc.* **2011**, 133, 13002. (b) Benkovics, T.; Guzei, I. A.; Yoon, T. P. *Angew. Chem., Int. Ed.* **2010**, 49, 9153.

(11) For a chiral phosphoric acid catalyzed example, see: Zhang, Z.; Antilla, J. C. *Angew. Chem., Int. Ed.* **2012**, 51, 11778.

(12) For recent research which found that the electrophilic indol-2-one could be controlled by the chiral nickel(II) catalyst, see: Zhang, H.; Hong, L.; Kang, H.; Wang, R. *J. Am. Chem. Soc.* **2013**, 135, 14098.

(13) For 3-halooxindoles as electrophiles examples, see: (a) Belmar, J.; Funk, R. L. *J. Am. Chem. Soc.* **2012**, 134, 16941. (b) Fuchs, J. R.; Funk, R. L. *Org. Lett.* **2005**, 7, 677. (c) Fuchs, J. R.; Funk, R. L. *J. Am. Chem. Soc.* **2004**, 126, 5068. (d) Hinman, R. L.; Bauman, C. P. *J. Org. Chem.* **1964**, 29, 2431. (e) Zuo, J.; Liao, Y.-H.; Zhang, X.; Yuan, W.-C. *J. Org. Chem.* **2012**, 77, 11325. (f) Liao, Y.-H.; Wu, Z.-J.; Han, W.-Y.; Zhang, X.-M.; Yuan, W.-C. *Chem.—Eur. J.* **2012**, 18, 8916. (g) Ma, S.; Han, X.; Krishnan, S.; Virgil, S. C.; Stoltz, B. M. *Angew. Chem., Int. Ed.* **2009**, 48, 8037. (h) Menozzi, C.; Dalko, P. I.; Cossy, J. *Chem. Commun.* **2006**, 42, 4638. (i) Xie, W.; Jiang, G.; Liu, H.; Hu, J.; Pan, X.; Zhang, H.; Wan, X.; Lai, Y.; Ma, D. *Angew. Chem., Int. Ed.* **2013**, 52, 12924.

■ NOTE ADDED AFTER ASAP PUBLICATION

References 5e–g contained errors in the version published ASAP April 11, 2014; the correct version reposted April 18, 2014.